High-Resolution Measurements of the Bands of Carbonyl Sulfide between 2510 and 3150 cm⁻¹

A. G. MAKI1

Molecular Physics Division, National Institute of Standards and Technology, Gaithersburg, Maryland 20899

J. S. WELLS

Time and Frequency Division, National Institute of Standards and Technology, Boulder, Colorado 80303

AND

J. B. BURKHOLDER

National Oceanic and Atmospheric Administration and Cooperative Institute for Research in Environmental Sciences, Boulder, Colorado 80303

The four strongest absorption bands of OCS in the region from 2500 to 3150 cm⁻¹ have been measured with FTS spectrometers with effective resolutions of about 0.010 cm⁻¹ or better. A total of 25 different vibrational transitions have been identified in the spectrum and have been analyzed to obtain improved band centers and rovibrational constants. Included in the bands identified are a few transitions due to the less abundant isotopomers, ¹⁶O ¹²C ³⁴S, ¹⁶O ¹²C ³³S, ¹⁶O ¹³C ³²S, and ¹⁸O ¹²C ³²S. Relative values are given for the transition moments of some of the overlapping bands. © 1991 Academic Press, Inc.

INTRODUCTION

In two fairly recent papers Fayt and co-workers (1, 2) have given the rovibrational constants for a great many vibrational states of the four most abundant isotopes of OCS. Many of those constants were based on recent laser-Stark, heterodyne, and FTS measurements. In the higher-frequency regions, however, the rovibrational constants were based on old data (3). The last published rovibrational analysis of the bands between 2700 and 3200 cm⁻¹ seem to be those of Maki *et al.* (4) reported in 1962. Quite recently, Dang-Nhu and Guelachvili (5) have reported the analysis of the ν_1 + ν_2 band² near 2575 cm⁻¹.

For the ¹⁶O¹³C³²S species Lahaye *et al.* (2) refer to more recent unpublished measurements. Measurements in this region are also given for the isotopomers ¹⁶O¹²C³⁴S and ¹⁸O¹²C³²S by Blanquet *et al.* (7) and by Fayt and Vandenhaute (8), respectively.

In this paper we report the measurement and analysis of the four strongest absorption

¹ Present address: 15012 24 Ave. S.E., Mill Creek, Washington 98012.

² The vibrational numbering system adopted by the IAU-IUPAP joint commission on spectroscopy (6) is used throughout this paper. Some papers use a notation that interchanges ν_1 and ν_3 .

bands of OCS in the region from 2500 to 3150 cm $^{-1}$. Actually, this region encompasses six transitions from the ground vibrational state, $00^{\circ}3-00^{\circ}0$, $11^{\circ}0-00^{\circ}0$, $02^{\circ}2-00^{\circ}0$, $10^{\circ}1-00^{\circ}0$, $04^{\circ}1-00^{\circ}0$, and $12^{\circ}0-00^{\circ}0$. Three other vibrational transitions from the ground state fall in this region but have transition moments that are too small to give observable absorption lines in our spectra.

Only three papers have reported measurements of the intensities of the bands covered in this paper. The two earliest papers, (9, 10) only give integrated intensities measured with spectrometers having resolution too poor to allow the measurement of individual line intensities. The most recent paper (5) gives line intensities for the 11^{10} – 00^{0} 0 band near 2575 cm⁻¹.

EXPERIMENTAL DETAILS

Two FTS spectrometers were used to make the measurements reported here. The bands between 2500 and 2970 cm⁻¹ were measured on the BOMEM ³ DA.002 at NIST in Gaithersburg, Maryland. These measurements all had an effective resolution (full width at half-height of weak lines) of 0.010 cm⁻¹. The bands between 2850 and 3150 cm⁻¹ were measured on a similar FTS spectrometer at NOAA in Boulder, Colorado. The NOAA measurements had an effective resolution of 0.004 cm⁻¹.

The NIST measurements used a single-pass absorption cell 0.75 m in length. A pressure of 1.3 kPa (9.9 Torr) was used for the measurements below 2800 cm⁻¹ and a pressure of .66 kPa (5.0 Torr) was used for the NIST measurements between 2850 and 2970 cm⁻¹. For the NOAA measurements a multipass absorption cell with a path length of 60 m was used with a pressure of 6 Pa (0.046 Torr) of OCS for measurements above 3000 cm⁻¹ and of 9 Pa (0.07 Torr) below 3000 cm⁻¹.

Each band system was measured separately and was calibrated by means of OCS absorption frequencies that had been determined from earlier hot band measurements at lower frequencies. For the region from 2500 to 2600 cm⁻¹ the calibration was provided by both the 00°3–00°0 and 11°0–00°0 bands which had been measured by heterodyne techniques (11, 12). The regions 2695–2765 cm⁻¹ and 2859–2990 cm⁻¹ were calibrated by the 02°2–00°0 and 04°1–00°0 bands, respectively (13). The 3025–3120 cm⁻¹ region was calibrated by the 12°0–00°0 band as determined by the measurements of Hunt *et al.* (14) after correcting to agree with more recent heterodyne measurements in the 2060 cm⁻¹ region (12). The exact calibration frequencies used in this work were not taken from the references cited above but were determined from a more recent fit of OCS frequency measurements that included a slightly different data base such as the inclusion of the recent far-infrared measurements of Vanek *et al.* (15).

ANALYSIS OF THE WAVENUMBER MEASUREMENTS

The transitions were assigned in most cases by calculating the transition frequencies from the constants given by Fayt *et al.* (1). For the less abundant isotopomers the

³ Brand names are used in this paper to identify experimental apparatus. Such use does not constitute an endorsement of products by NIST.

transitions were calculated or estimated from various other works of Fayt and coworkers (2, 7, 8).

With the exception of a few transitions as noted later, the measurements were analyzed by means of the equations given by Maki *et al.* (13). The analysis takes into account the effects of *l*-type resonance but does not include any specific allowance for Fermi resonance.

It is important to note that the band centers, given in Table I, are defined by the equations

$$E^{0} = G + B_{v}J(J+1) - D_{v}[J(J+1) - l^{2}]^{2} + H_{v}[J(J+1) - l^{2}]^{3} + L_{v}[J(J+1) - l^{2}]^{4} + \cdots$$
 (1)

and

$$\nu_0 = G' - G''. \tag{2}$$

Some authors use a somewhat different definition for the band centers, ν_0 . In Eq. (1) E^0 denotes the energy level with the effect of *l*-type resonance removed, sometimes

TABLE I

Band Centers Determined in This Analysis.

Transition	v _o this work (cm ⁻¹)	v _o other work (cm ⁻¹)	[Ref]	rms dev. (cm ⁻¹)	J ^a max
¹⁶ O ¹² C ³² S					
00 ⁰ 3-00 ⁰ 0*b	2555.991 22(1	3)° 2555.990 91(66)	[1]	0.0004	75
01 ¹ 3-01 ¹ 0	2536.670 98(5	1) 2536.669 65(162	(1)	0.0013	45
$00^{0}4-00^{0}1$	2535.002 20(5	9) 2535.001 96(212	(1)	0.0011	44
11 ¹ 0-00 ⁰ 0*	2575.307 59(8) 2575.307 74(23)	[1]	0.0005	74
02 ⁰ 2-00 ⁰ 0*	2731.399 12(2	(4) 2731.399 08(56)	[1]	0.0006	77
03 ¹ 2-01 ¹ 0	2724.838 52(2	(5) 2724.838 95(112	2) [1]	0.0008	60
$02^{0}3-00^{0}1$	2705.512 88(9	7) 2705.519 52(159) [1]	0.0017	40
$10^{0}1 - 00^{0}0$	2918.105 14(3) 2918.105 53(40)	[1]	0.0004	99
11 ¹ 1-01 ¹ 0	2903.717 64((1) 2903.718 11(31)	[1]	0.0003	78
10 ⁰ 2-00 ⁰ 1	2909.530 15(2	1) 2909.539 61(284	1) [1]	0.0004	50
12 ⁰ 1-02 ⁰ 0	2890.385 31(3			0.0004	57
12 ² 1-02 ² 0	2890.008 25(2		[1]	0.0003	56
11 ¹ 2-01 ¹ 1	2893.865 89(52) 2893.868 31(22 ⁹) [1]	0.0007	48
$04^{0}1-00^{0}0*$	2937.146 85(2	2937.148 24(33)	[1]	0.0005	82
$04^{0}2-00^{0}1$	2903.858 68(9	2903.852 43(28)	L) [1]	0.0005	50
12 ⁰ 0-00 ⁰ 0	3095.554 42(.1) 3095.554 71(42)	[1]	0.0003	68
13 ¹ 0-01 ¹ 0 180 ¹² C ³² S	3094.923 27([1]	0.0004	50
10 ⁰ 1-00 ⁰ 0 160 ¹² C ³³ S	2860.687 37(70) 2860.685(16)	[8]	0.0007	50
$10^{0}1-00^{0}0$ $^{16}O^{12}C^{34}S$	2911.905 77(27)		0.0005	67
00 ⁰ 3-00 ⁰ 0	2523.003 11(8	33) 2523.000 76(190) [2]	0.0012	39
02 ⁰ 2-00 ⁰ 0	2708.313 89	175)		0.0018	33
10 ⁰ 1-00 ⁰ 0	2906.045 61(17) 2906.054 7(48)	[7]	0.0003	75
1111-0110	2891.991 87(16)	-	0.0007	50
12 ⁰ 0-00 ⁰ 0	3092.797 37(70)		0.0007	36
10 ⁰ 1-00 ⁰ 0	2861.049 85(31) 2861.050(10)	[2]	0.0005	62

a) J_{max} refers to the highest value of J that contributes to the fit for the upper state energy levels. In some cases the present measurements may have terminated at lower values of J.

b) For transitions indicated by an asterisk (*) the band center was determined from heterodyne measurements reported earlier.
 c) The uncertainty (twice the standard error) in the last digits is given in parentheses.

called the deperturbed energy level. Our analysis treats the l-doubling as an effect of the l-type resonance. The l-doubling constant, q, has the form

$$q = q_{v} - q_{vJ}J(J+1) + q_{vJJ}[J(J+1)]^{2}.$$
 (3)

Table I gives the band centers, the rms deviation of the fits, and the maximum J value included in the fit for each vibrational transition. Tables II and III give the rovibrational constants used or determined for each vibrational energy level involved in these measurements. The least-squares fits used in this analysis included a large body of infrared and microwave data to better define the constants. The uncertainties given in the tables are the statistical uncertainties given by the least-squares fit and do not include any allowance for systematic errors that may be hidden in the data. We believe that the main effect of such systematic errors will be an underestimate of the true error in the band centers. Probably a more realistic estimate of the accuracy of the band centers would be given by adding $0.0006~\rm cm^{-1}$ to the uncertainties of the band centers given in Table I, except for those bands for which there are heterodyne measurements.

In the tables the separation of the vibrational energy levels is not given for the levels that differ only in the value of l. Those separations have been given by Fayt $et\ al.\ (1)$ with sufficient accuracy to calculate the effect of l-type resonance. Some of our earlier papers give small corrections to those separations $(13,\ 15)$.

For the $02^{\circ}2-00^{\circ}0$ and $12^{\circ}0-00^{\circ}0$ transitions of $^{16}O^{12}C^{34}S$, we had no accurate estimate of the location of the l=2 levels so it was not possible to include the effect of l-type resonance in the analysis. Instead, the transitions were fit to the usual power series in J(J+1) as given in Eq. (1). This accounts for the unusually small values of D_v for the upper states involved in those transitions.

RELATIVE INTENSITY MEASUREMENTS

Although the current spectra were not intended to be used for intensity measurements, the nearly complete absence of intensity data for this region prompted us to measure the relative intensities of a few lines. The results of the present measurements are given in Table IV.

The relative intensity measurements are based on only crude estimates of the relative intensities of adjacent lines that are weak and of similar intensity. The relative intensities were verified by comparing calculated spectra with the measured spectrum. For those hot band and isotopic transitions that are given in Table I, but not included in Table IV, the calculated spectrum was in good agreement with the measured spectrum on the assumption that the transition moment is the same as the ground state transition for the most abundant isotopic species.

The values given in Table IV are based on the following equation for the integrated intensity, S, of individual rovibrational absorption lines:

$$S = \exp(-E''/kT)[1 - \exp(-\nu/0.69504T)]$$

$$\times [N_{\rm i}/Q_{\rm v}Q_{\rm R}]\nu C |\mu(v'-v'')|^2 S_{\rm v}^2 S_{\rm R}^2 F, \quad (4)$$

where C is a proportionality constant that includes $8\pi^3/(3hc)$ and other factors, N_i is the concentration of the isotopic species under consideration, Q_v and Q_R are the vi-

Vib. State	•	В		D _v ×10 ⁸	н _у ×10 ¹⁴	q _v ×10 ⁴	q _{vJ} ×10 ¹⁰
0000	0.202	856	740 80(83)ª	4.340 64(25)	-0.329(30)		
0110			834 8(21)	4.411 48(31)	-0.260(38)	2.121 938 68(53)	1.424 13(102)b
0001	0.202	251	831 6(60)	4,433 50(36)	0.045(45)		
02 ⁰ 0	0.203	480	484 7(123)	4.419 64(63)	-0.712(73)	2.086 287(47)	0.659(35)
02 ² 0	0.203	559	482 1(89)	4.483 28(93)	-0.135(80)		
011	0.202	657	042(22)	4.542 71(63)	[0.1]	2.285 20(29)	3.593(114)
00 ⁰ 3	0.201	006	219(100)	4.647 59(514)	1.60(70)		
1110	0.202	015	427 (45)	4.473 80(210)	2.85(28)	2.155 33(27)	3.764(70)
0202	0.202	414	701 (226)	4.707(108)	[-0.3]	2.365(100)	2.76(137)
0222	[0.202	523	2]	[4.804 7]	[0.1]		
10 ⁰ 1 ^d	0.201	102	978(109)	5.071 26(684)	54.36(148)		
04 ⁰ 1 ^e	0.203	436	473 (356)	3.943 2(254)	-52.44(633)	2.127 48(95)	-1.168(169)
0421	0.203	569	074(81)	4.668 2(13)	[-0.3]		
	[0.203	799	95]	[4.34]	[-0.3]		
01 3	0.201	510	88 (143)	4.824(80)	[2.2]	2.596 1(160)	3.2(118)
12 ⁰ 0	0.202	311	240(147)	4.500 9(56)	[2.2]	2.130 1(101)	6.6(28)
12 ² 0	0.202	382	585 (146)	4.531 9(48)	[2.2]		
03 <u>1</u> 2	0.202	744	75 (41)	4.771 1(127)	[-0.2416]	2.302 77(227)	2.93(89)
0332	[0.202	936	096]	[4.776 6]	[-0.2416]		
0004	0.200	363	45(170)	4.792(94)	[2.5]		
11 ¹ 1	0,201	515	903(248)	4.755 4(110)	8.57(130)	2.373 71(134)	8.55(32)
0203			71(330)	4.989(216)	[2.397]	[2.42]	[0.6587]
02 ² 3	(0.201	981	318]	[4.983 28]	[2.397]		
13 ¹ 0 13 ³ 0 04 ⁰ 2 ^f	0.202	617	950(628)	4.569(66)	6.7(186)	2.100 35(152)	2.06(86)
1350	[0.202	745	521)	[4.41]	[-1.948]		
04021			23(986)	15.06(342) -	5459.(5096)		
10 ⁰ 2 ^g					3965.(687)		
1201			303 (531)	4.745 3(178)	[6.00]	[2.255 63]	[3.7]
1221			654 (434)	4.782 4(151)	[6.00]	•	•
11 ¹ 2	0.201	069	28(118)	6.539 7(543)	[240.0]	2.709 1(142)	47.6(84)

TABLE II Rotational Constants (in cm⁻¹) Used to Fit the Present Measurements for ¹⁶O ¹²C ³²S

b١

brational and rotational partition functions, respectively, T is the temperature, E'' is the lower state energy, and ν is the wavenumber of the transition (in cm⁻¹).

When ratios of line intensities are compared, as in this work, most of the terms in Eq. (4) cancel or are easily calculated. The principal terms that do not cancel are $|\mu(v'-v'')|$, the transition moment or dipole derivative, S_R , the rotational strength factor or direction cosine matrix element for rotation, S_{v} , the vibrational strength factor, and F, the Herman–Wallis factor.

In this paper we have used a Herman-Wallis factor that has the form

$$F = 1 + a_1 m + a_2 J'(J' + 1), (5)$$

where m has the usual meaning of J' for R-branch transitions and -J'' for P-branch transitions. For the transitions involving $\Delta v_2 = 1$, a nonzero value for a_1 is usually required, as found by Dang-Nhu and Guelachvili (5). For the 10^o2-00^o1 and 04^o2-00°1 transitions a resonance perturbation requires that we use a nonzero value for the a_2 term.

The rotational strength factors, S_R , were given by the usual equations. For $\Delta l = 0$, the $\Delta J = 0$ transitions have

$$S_{\rm R} = [(2J+1)l^2/J(J+1)]^{1/2},$$
 (6)

The uncertainty in the last digits (twice the estimated standard error) is given in parentheses

d)

e)

The parentheses. Also included in the fit was $q_{y,j}=0.574(44)\times10^{-14}$. The values enclosed in square brackets were fixed during the analysis. An additional term, L=-1.985(95)×10⁻¹⁷, was needed in the analysis. An additional term, L=1.71(50)×10⁻¹⁷, was needed in the analysis. Additional terms needed in the analysis were: L=0.16(366)×10⁻¹⁴, M=-1.03(125)×10⁻¹⁷, and N=4.26(161)×10⁻²¹. f)

Additional terms needed in the analysis were: L=2.901(326) \times 10⁻¹⁴ and M=-8.41(54) \times 10⁻¹⁸. See text for discussion of analysis beyond .T=50 q)

and the $\Delta J = \pm 1$ transitions have

$$S_{R} = [(|m|^{2} - l^{2})/|m|]^{1/2}.$$
 (7)

For $\Delta l = \pm 1$ and $\Delta J = 0$

$$S_{\rm R}^2 = \frac{1}{2}(2J+1)(J+l+1)(J-l)/J(J+1). \tag{8}$$

For $\Delta l = +1$ and $\Delta J = +1$

$$S_{\rm R}^2 = \frac{1}{2}(J+l)(J+l+1)/J \tag{9}$$

while for $\Delta l = +1$ and $\Delta J = -1$

$$S_{\rm R}^2 = \frac{1}{2}(J - l)(J - l - 1)/J,\tag{10}$$

where J is the larger of J' and J'' but l is always l''.

The vibrational strength factors were slightly different for even and odd values of Δv_2 . For simplicity the strength factor was broken into two factors such that

$$S_{\rm v}^2 = S_{13}^2 S_2^2$$

with

$$S_{13}^2 = (v_1 + \Delta v_1)!(v_3 + \Delta v_3)!/(v_1!v_3!\Delta v_1!\Delta v_3!)$$
(11)

and for Δv_2 even (or zero)

$$S_{2}^{2} = \left[\frac{1}{2}(v_{2} + l + \Delta v_{2})\right]! \left[\frac{1}{2}(v_{2} - l + \Delta v_{2})\right]! / \left\{\left[\frac{1}{2}(v_{2} + l)\right]! \left[\frac{1}{2}(v_{2} - l)\right]! \left[\left(\frac{1}{2}\Delta v_{2}\right)!\right]^{2}\right\}, \quad (12)$$

TABLE III Rotational Constants (in cm⁻¹) Used to Fit the Present Measurements for Different Isotopes of OCS

Vib. State		B_v		D _v ×10 ⁸	$H_v \times 10^{14}$
¹⁸ O ¹² C ³² S					
0000	[0.190	293	541 5]ª	[3.778 18]	[-0.300]
10 ⁰ 1 160 ¹² C ³³ S	0.188	619	16(148) ^b	4.041(60)	[30.0]
0000	0.200	302	432(17)	4.237 48(159)	-0.314(433)
10 ⁰ 1 160 ¹² C ³⁴ S	0.198	574	71 (66)	4.948(38)	36.7(60)
0000	0.197	898	035 1(67)	4.141 19(125)	-0.351(184)
01 ¹ 0°	0.198	242	548 7(107)	4.208 25(86)	-0.319(91)
00 ⁰ 3	0.196	093	46(293)	4.429(206)	[1.151]
02 ⁰ 2	0.197	464	68 (760)	3.423(672)	[22.5]
10 ⁰ 1	0.196	196	851(318)	4.916 1(155)	41.78(204)
12 ⁰ 0	0.197	377	02 (92)	[3.8]	[8.8]
11 ¹ 1 ^d 160 ¹³ C ³² S	0.196	588	43 (99)	4.556(40)	[10.0]
0000	0.202	204	013 7(99)	4.328 8(21)	-0.269(379)
1001			85 (98)	7.161 4(697)	232.9(129)

a) The values enclosed in square brackets were fixed during the analysis.

b) The uncertainty in the last digits (twice the estimated standard error) is given in parentheses.

scandaru error) is given in parentheses. c) The ℓ -doubling terms were $q_v{=}2.024~338~01(78)\times10^{-4}$ and $q_{v_J}{=}1.3084(74)\times10^{-10}.$ d) The ℓ -doubling terms were $q_v{=}2.2610(130)\times10^{-4}$ and $q_{v_J}{=}14.9\,(66)\times10^{-10}.$

TABLE IV Measurements of Relative Transition Moments

$$\frac{|\mu(00^{0}3-00^{0}0)|}{|\mu(11^{1}0-00^{0}0)|} = 1.43(14)^{a} \qquad |\mu(11^{1}0-00^{0}0)|^{b} = 0.00257 \text{ debye}^{c}$$

$$\frac{|\mu(00^{0}3-00^{0}0)|}{|\mu(00^{1}3-01^{1}0)|} = 1.29(13)$$

$$\frac{|\mu(00^{1}3-01^{1}0)|}{|\mu(01^{1}3-01^{1}0)|} = 1.026(100)$$

$$\frac{|\mu(10^{0}1-00^{0}0)|}{|\mu(04^{0}1-00^{0}0)|} = 9.444(944)$$

$$\frac{|\mu(10^{0}1-00^{0}0)|}{|\mu(05^{1}1-01^{1}0)|} = 1.44(14)^{d}$$

$$\frac{|\mu(10^{0}1-00^{0}0)|}{|\mu(10^{0}2-00^{0}1)|} = 1.097(110)^{e}$$

$$\frac{|\mu(10^{0}1-00^{0}0)|}{|\mu(04^{0}2-00^{0}1)|} = 2.33(23)^{f}$$

- a) The uncertainty of these measurements is believed to be
- about 10% of the given values and is given in parentheses.
 b) The transition moment for the 110-00% transition was adjusted to agree with the measurements of Dang-Nhu and Guelachvili (5), including the term a, = 0.00463.
 c) 1 debye = 3.336×10⁻³⁰ C m.
- d) The transition moments for hot bands do not include the vibrational dependence terms that come from Eqs. (11, 12, and 13).
- e) A Herman-Wallis term of $a_2 = -0.00034$ is also needed for the line intensities of the 10^02-00^01 band.
- f) A Herman-Wallis term of $a_2 = 0.00105$ is also needed for the line intensities of the 04^02-00^01 band.

while for Δv_2 odd

$$S_{2}^{2} = \left[\frac{1}{2}(v_{2} + l + \Delta v_{2} - 1)\right]! \left[\frac{1}{2}(v_{2} - l + \Delta v_{2} - 1)\right]! (v_{2} + \Delta l * l + \Delta v_{2} + 1) / \left\{ \left[\frac{1}{2}(v_{2} + l)\right]! \left[\frac{1}{2}(v_{2} - l)\right]! \left\{ \left[\frac{1}{2}(\Delta v_{2} - 1)\right]! \right\}^{2} (\Delta v_{2} + 1) \right\}.$$
 (13)

Equations (12) and (13) only apply if $|\Delta l| \le 1$, otherwise $S_2 = 0$. In Eqs. (11), (12), and (13) the smaller of v' and v'' is used for v, $\Delta v = |v' - v''|$, l = l'', and Δl = l' - l''.

Equation (11) was derived from the properties of harmonic oscillator wave functions such as can be found, among other places, in Appendix III of Wilson et al. (16). Equations (12) and (13) were derived from the properties of the two-dimensional harmonic oscillator wave functions given by Moffitt and Liehr (17).

Note that S_{ν}^2 is normalized so that transitions from the ground state always have $S_{\rm v}^2 = 1$. For some hot bands $S_{\rm v}$ may be 1, but for certain hot bands, such as $10^{0}2$ - $00^{\circ}1$, $S_{v}^{2} = 2$, while for the hot band $00^{\circ}4-00^{\circ}1$, $S_{v}^{2} = 4$. Other authors sometimes include S_v in the dipole derivative term, $|\mu|$, in which case the dipole derivative for certain hot bands will be very different from that for the ground state transitions.

THE
$$\Delta v_1 = -1$$
, $\Delta v_2 = 4$ RESONANCE

For some of the transitions measured in this work there is a significant resonance coupling levels of the type (v_1, v_2, v_3, l) and $(v_1 - 1, v_2 + 4, v_3, l)$. This resonance has been recognized (1, 3, 14, 18) as contributing to unusually large values for the centrifugal distortion constants, D, H, L, etc., for levels with either $v_1 \neq 0$ or $v_2 \geq 4$. The levels $10^{0}1, 04^{0}1$, and $11^{1}2$ are good examples of how this weak resonance affects the D_v and H_v terms.

Fayt (18) has shown that the $10^{0}2$ and $04^{0}2$ levels present a case where this resonance is particularly significant since the levels are only 5.7 cm⁻¹ apart and the difference in their B values is such that their unperturbed energy manifolds would cross at $\sim J$ = 39. Fayt has also found that the $04^{2}2$ level is only 8.8 cm⁻¹ below the $10^{0}2$ level and, because of the difference in B values, their manifolds cross at $\sim J$ = 55. Normally, l-type resonance couples the $04^{2}2$ and $04^{0}2$ levels. Consequently, it will couple the $04^{2}2$ and $10^{0}2$ levels because the $\Delta v_{1} = -1$, $\Delta v_{2} = 4$ resonance mixes the wavefunctions of the $10^{0}2$ and $04^{0}2$ levels. This three level resonance has been carefully explained and analyzed by Fayt (3, 18) who had a more complete set of transitions from the ground state than we have from the $00^{0}1$ state.

In this work we have found transitions up to J=55 for the two highest-frequency rotational manifolds which we have labeled $10^{0}2-00^{0}1$ and $04^{0}2-00^{0}1$ according to the vibrational character of the lowest J levels. We have also found a fragment of the lowest energy levels of this triad between J=54 and 70 which at the highest J levels also becomes primarily a $10^{0}2-00^{0}1$ transition. This latter fragment was not included in the fit that gave the constants in Tables I and II because the two avoided crossings make it impossible to fit both the high J and low J transitions to a single power series, such as Eq. (1).

The intensity of the observed transitions comes almost entirely from the transition moment of the $10^{0}2-00^{0}1$ transition. For a quantitative description of the intensities of the transitions for the $10^{0}2-00^{0}1$ and or $04^{0}2-00^{0}1$ bands it would be necessary to take into account the mixing of the wavefunctions of the three levels involved in the interaction. An examination of the observed intensities and the mixing implied by the energy level diagram (Fig. 1 in Fayt (18)) leads us to believe that the intensities are in very good agreement with Fayt's analysis.

We have taken a more qualitative approach and have assigned transition moments to the low J transitions and given approximate values for effective Herman-Wallistype constants. These are given in Table IV. The constants given in Tables I, II, and IV for these two interacting levels should not be applied to levels for J > 50.

One of the results of this resonance is the enhancement of the intensity of the 04^02 – 00^01 transitions beyond what one would expect from the intensity of the 04^01 – 00^00 transitions. The 04^02 – 00^01 and 10^02 – 00^01 transitions have nearly equal intensities because the wavefunctions of the upper states are so strongly mixed. The sum of the intensities of the two transitions is approximately equal to what the sum of the intensities would have been if there had been no coupling of the states.

ACKNOWLEDGMENT

This work was supported in part by the NASA Office of Upper Atmospheric Research.

RECEIVED: December 19, 1990

REFERENCES

- I. A. FAYT, R. VANDENHAUTE, AND J. G. LAHAYE, J. Mol. Spectrosc. 119, 233-266 (1986).
- 2. J. -G. Lahaye, R. Vandenhaute, and A. Fayt, J. Mol. Spectrosc. 123, 48-83 (1987).
- 3. A. FAYT, Ann. Soc. Sci. Bruxelles 84, 69-106 (1970).
- 4. A. G. MAKI, E. K. PLYLER, AND E. D. TIDWELL, J. Res. Natl. Bur. Strand. Sect. A 66, 163-167 (1962)
- 5. M. DANG-NHU AND G. GUELACHVILI, Mol. Phys. 58, 535-540 (1986).
- 6. R. S. MULLIKEN, J. Chem. Phys. 23, 1997-2011 (1955).
- 7. G. BLANQUET, J. WALRAND, C. P. COURTOY, AND A. FAYT, J. Mol. Spectrosc. 81, 473-479 (1980).
- 8. A. FAYT AND R. VANDENHAUTE, Mol. Phys. 31, 1861-1873 (1976).
- 9. A. FOORD AND D. H. WHIFFEN, Mol. Phys. 26, 959-968 (1973).
- 10. R. H. KAGANN, J. Mol. Spectrosc. 94, 192-198 (1982).
- 11. J. S. WELLS, M. D. VANEK, AND A. G. MAKI, J. Mol. Spectrosc. 135, 84-88 (1989).
- 12. J. S. Wells, M. Schneider, and A. G. Maki, J. Mol. Spectrosc. 140, 170-176 (1990).
- 13. A. G. Maki, Wm. B. Olson, J. S. Wells, and M. D. Vanek, J. Mol. Spectrosc. 130, 69-80 (1988).
- N. Hunt, S. C. Foster, J. W. C. Johns, and A. R. W. McKellar, J. Mol. Spectrosc. 111, 42–53 (1985).
- 15. M. D. VANEK, D. A. JENNINGS, J. S. WELLS, AND A. G. MAKI, J. Mol. Spectrosc. 138, 79–83 (1989).
- E. B. WILSON, J. C. DECIUS, AND P. C. CROSS, "Molecular Vibrations—The Theory of Infrared and Raman Vibrational Spectra," McGraw-Hill, New York, 1955.
- 17. W. MOFFITT AND A. D. LIEHR, Phys. Rev. 106, 1195-1200 (1957).
- 18. A. FAYT, Ann. Soc. Sci. Bruxelles 82, 101-112 (1968).